

AD-A253 384



92-18841



UNCLASSIFIED

NSN 7540-01-280-5500

TABLE OF CONTENTS

	<u>Page</u>
LIST OF FIGURES	vi
LIST OF TABLES	vii
1. INTRODUCTION	1
2. OVERVIEW OF PROCEDURE	3
3. UNCERTAINTY IN ESTIMATES OF β IN SEAWATER	5
4. RESULTS	9
4.1 PLOTS OF c , ρ , β , AND Λ	9
4.2 POLYNOMIAL EXPRESSIONS FOR β AND Λ	12
4.3 DISCREPANCY WITH THE TABULATED VALUES OF ENDO [1]	16
5. SUMMARY	17
ACKNOWLEDGMENTS	19
REFERENCES	21

LIST OF FIGURES

<u>Figure</u>	<u>Page</u>
3.1 Plots of the absolute difference in β versus temperature for absolute pressures of 1, 100, 200, 300...800 bar caused by using different empirical relations for c in term 1.	6
3.2 Plots of the absolute difference in β versus temperature for absolute pressures of 1, 100, 200, 300...800 bar caused by using different empirical relations for c in term 2.	6
3.3 Plots of the absolute difference in β versus temperature for absolute pressures of 1, 100, 200, 300...800 bar caused by using different empirical relations for ρ and α in term 1.	6
3.4 Plots of the absolute difference in β versus temperature for absolute pressures of 1, 100, 200, 300...800 bar caused by using different empirical relations for ρ and α in term 2.	6
4.1 Plots of sound speed versus temperature for various absolute pressures (bar) in fresh water.	10
4.2 Plots of density versus temperature for various absolute pressures (bar) in fresh water.	10
4.3 Plots of β versus temperature for various absolute pressures (bar) in fresh water.	10
4.4 Plots of Λ versus temperature for various absolute pressures (bar) in fresh water.	10
4.5 Plots of sound speed versus temperature for various absolute pressures (bar) in seawater with a salinity of 35 ppt. [10]	11
4.6 Plots of density versus temperature for various absolute pressures (bar) in seawater with a salinity of 35 ppt. [12]	11

4.7	Plots of β versus temperature for various absolute pressures (bar) in seawater with a salinity of 35 ppt.	11
4.8	Plots of Λ versus temperature for various absolute pressures (bar) in seawater with a salinity of 35 ppt.	11

LIST OF TABLES

<u>Table</u>		<u>Page</u>
4.1	Coefficients in polynomial expression for β	13
4.2	Coefficients in polynomial expression for Λ	15

DTIC QUALITY INSPECTED 2

Accession For	
NTIS GRA&I	<input checked="" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification _____	
By _____	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
A-1	

1. INTRODUCTION

In the study of nonlinear acoustic propagation through an inhomogeneous fluid such as seawater, it is helpful to have explicit relations for the acoustic properties of the fluid as a function of temperature, pressure, and composition. For both fresh water and seawater, several empirical relations give the sound speed c and the density ρ as functions of temperature, pressure, and, in the case of seawater, salinity S . We are, however, unaware of any similar relations for the coefficient of nonlinearity β . Endo [1] has tabulated β for seawater over a broad range of temperatures and pressures for salinities of 25, 30, 35, and 40 parts per thousand (ppt). For computer implementations of nonlinear propagation through seawater, however, relations with explicit temperature, pressure, and salinity dependence are preferable over tables of values. Endo's tabular values were useful in checking our calculations.

In this work three relations for β are developed: one for fresh water as a function of temperature and pressure, one for seawater with a salinity of 35 ppt as a function of temperature and pressure, and one for seawater as a function of temperature, pressure, and salinity. Also developed is a similar set of relations for the term $\Lambda \equiv \beta/(\rho c^5)^{1/2}$.

The reason for developing separate relations for Λ is that it arises as an important variable in nonlinear geometrical acoustics [2,3]. Under the assumptions of nonlinear geometrical acoustics, the equation governing the propagation of a progressive finite-amplitude acoustic wave in an inhomogeneous fluid may be converted to the equation governing the propagation of a similar plane wave in a homogeneous fluid. The conversions required are the following:

$$\bar{p} \equiv p \left(\frac{A}{A_0} \right)^{1/2} \left(\frac{\rho c}{\rho_0 c_0} \right)^{-1/2}, \quad (1.1)$$

$$\bar{x} \equiv \int_{\ell_0}^{\ell} \frac{\Lambda}{\Lambda_0} \left(\frac{A}{A_0} \right)^{-1/2} d\ell', \quad (1.2)$$

where p is the acoustic pressure, \bar{x} and \bar{p} are, respectively, the equivalent plane wave propagation distance and pressure, A is the ray tube area, and ℓ is the ray path length. The subscript $_0$ indicates conditions at the source point. Equation (1.2) makes clear why an explicit relation for Λ as a function of temperature, pressure, and salinity is helpful.

2. OVERVIEW OF PROCEDURE

The coefficient of nonlinearity is defined as follows: $\beta = B/2A+1$, where A and B are the first and second coefficients of the Taylor series expansion of the total pressure P in terms of the relative density change. (In fresh water at room temperature and atmospheric pressure, β has a value of approximately 3.5.) Rudnick [4] has shown that B/A may be expressed in terms of quantities that are easily related to physical data. Use of the expression for B/A as shown in the paper by Beyer [5] gives

$$\beta = \rho c \left(\frac{\partial c}{\partial P} \right)_T + \frac{c\alpha T}{C_P} \left(\frac{\partial c}{\partial T} \right)_P + 1 \quad , \quad (2.1)$$

where the subscripts indicate constant conditions, and T , α , and C_P are, respectively, the absolute temperature, the coefficient of thermal expansion, and the specific heat at constant pressure. Note that the first and second terms in Eq. (2.1) contain different partial derivatives of the sound speed $c(P, T)$ evaluated at constant composition. The term that contains $(\partial c/\partial P)_T$ is hereinafter referred to as “term 1”, whereas the term that contains $(\partial c/\partial T)_P$ is hereinafter referred to as “term 2”. Rudnick noted that, while all parts make a significant contribution to β , term 1 is numerically larger than term 2 in all common fluids.

In our evaluation of β , published empirical relations were used as much as possible. When a derivative of an acoustical property was required, the derivative was obtained by differentiating the published empirical relation analytically, then evaluating the result numerically.

To the best of our knowledge, there is only one set of empirical relations for the acoustical properties of fresh water that is valid over a wide range of temperatures (0–100°C) and pressures (1–1000 bars absolute). The relations used for the sound speed and density of fresh water are from, respectively, Chen and Millero [6] and Chen *et al.* [7]. The estimates of the coefficient of thermal expansion were obtained using ρ from Chen *et al.* [7] in the expression $-\frac{1}{\rho} \left(\frac{\partial \rho}{\partial T} \right)_P$. For C_P , on the other hand, we used a polynomial fit to a fresh water C_P data set [8].

The situation is slightly different for seawater: at least two published empirical relations that are valid over a wide range of temperatures (0–40°C) and pressures (1–800 bars absolute) exist for both the sound speed and the density. For the sound speed the choice is between the relation developed by Lovett [9] and that developed by Chen and Millero [10]. For the density the choice is between the expressions

developed by Gebhart and Mollendorf [11] and those from Chen and Millero [12]. The choice for the coefficient of thermal expansion is somewhat dependent on the choice made for the density. The choice is either the evaluation of $-\frac{1}{\rho} \left(\frac{\partial \rho}{\partial T} \right)_{P,S}$ using ρ from Gebhart and Mollendorf [11] or the explicit relation for α given by Chen and Millero [12]. Our estimate of the specific heat at constant pressure in seawater C_P was obtained by summing the estimate of C_P at one atmosphere, which was calculated using a relation from Millero *et al.* [13], with the contribution from the isothermal integration of [14]

$$-T \int \frac{1}{\rho} \left(\alpha^2 + \left(\frac{\partial \alpha}{\partial T} \right) \right) dP \quad . \quad (2.2)$$

The estimates of α and ρ for Eq. (2.2) were obtained from the appropriate aforementioned references.

3. UNCERTAINTY IN ESTIMATES OF β IN SEAWATER

To help decide which empirical relations to use for c , ρ , and α in seawater and, more importantly, to estimate the uncertainty in the calculation of β , we numerically implemented all of the aforementioned empirical relations. The absolute difference in β due to using two different empirical relations was calculated for one property at a time. The difference in β caused by terms 1 and 2 was calculated separately to see which term in β is more sensitive.

Term 1, the term in β that contains $(\partial c / \partial P)_T$, was found to be the largest source of uncertainty. Estimates of the uncertainty were obtained by calculating term 1 using Lovett's relation for c [9] while holding all other relations constant. (For ρ and α , the previously referenced Chen and Millero relations were used.) This result was then subtracted from a calculation of term 1 obtained using Chen and Millero's relation for c [10]. The difference was calculated over a realistic range of temperatures and pressures, that is, a range of temperatures and pressures that might be found in an ocean: a temperature range of -4 – 40° for pressures of 1 and 100 bars absolute, but a temperature range of only -4 – 10° for pressures of 200, 300, . . . , 800 bars absolute. Shown in Fig. 3.1 is a plot of the difference in term 1 for salinities of 30 and 40 ppt. (Solid lines are for salinity of 30 ppt, dashed for 40 ppt.) For these salinities, the difference is always less than 0.1, typically 0.05. For a salinity of 35 ppt (not shown), the absolute difference is always less than 0.05 and typically less than 0.025. For contrast, Fig. 3.2 shows a plot of the corresponding difference in term 2, which was calculated in an identical fashion. The magnitude of these differences is much smaller, typically less than 0.0025.

The uncertainties in β that stem from the different empirical relations for ρ and α were examined in the same manner as those for c . Both terms 1 and 2 were evaluated twice, once using the density relation from Gebhart and Mollendorf [11] and once using Chen and Millero's relation [12]. Because α is closely related to ρ , it was varied simultaneously. (In all cases, the relation used for c was that of Chen and Millero [10].) Plots of the differences generated by terms 1 and 2 are shown, respectively, in Figs. 3.3 and 3.4. The largest differences in β that result are in term 2, and the differences are always less than 0.01.

Finally, a check on the pressure correction term in C_p , Eq. (2.2), was made at zero salinity. Between 1 and 1000 bar, the pressure increment in C_p given by Eq. (2.2) deviated by around 10% from the values in Ref 8. Nevertheless, the corresponding uncertainty in β , through term 2, is only 1 part in 10^4 .

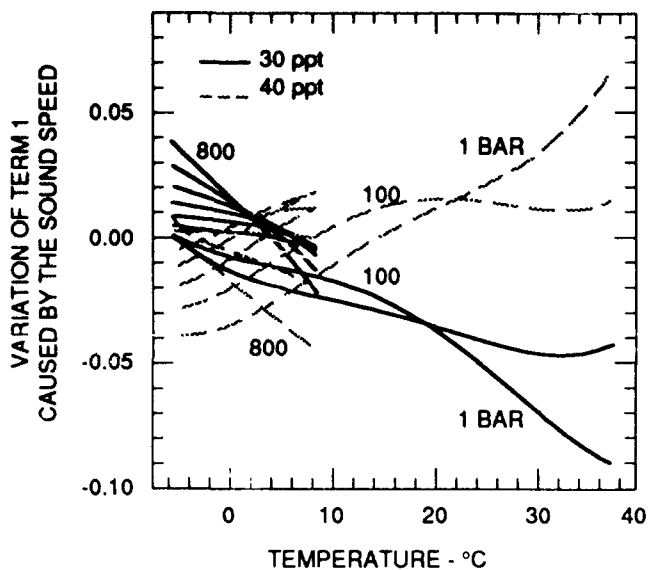


FIGURE 3.1
PLOTS OF THE ABSOLUTE DIFFERENCE IN β versus TEMPERATURE FOR ABSOLUTE PRESSURES OF 1, 100, 200, 300...800 BAR CAUSED BY USING DIFFERENT EMPIRICAL RELATIONS FOR c IN TERM 1

AS-91-649

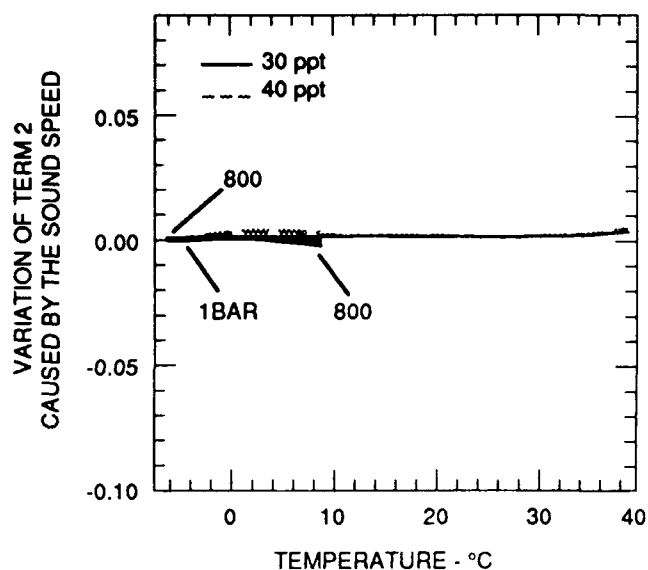


FIGURE 3.2
PLOTS OF THE ABSOLUTE DIFFERENCE IN β versus TEMPERATURE FOR ABSOLUTE PRESSURES OF 1, 100, 200, 300...800 BAR CAUSED BY USING DIFFERENT EMPIRICAL RELATIONS FOR c IN TERM 2

AS-91-650

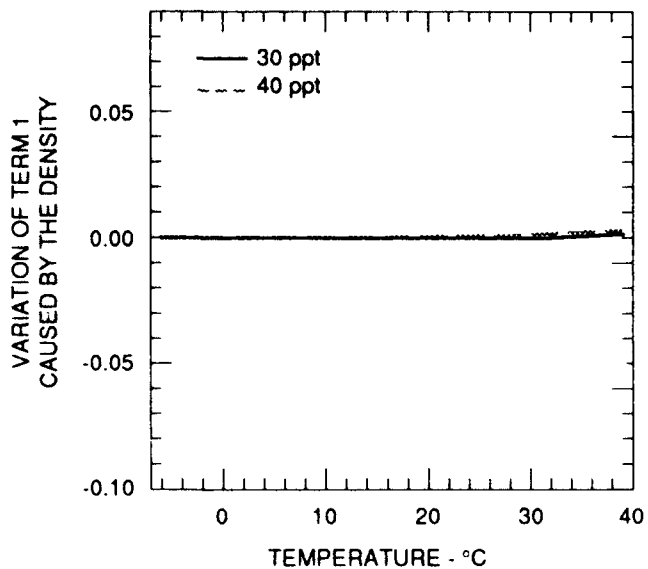


FIGURE 3.3
PLOTS OF THE ABSOLUTE DIFFERENCE IN β versus TEMPERATURE FOR ABSOLUTE PRESSURES OF 1, 100, 200, 300...800 BAR CAUSED BY USING DIFFERENT EMPIRICAL RELATIONS FOR ρ AND α IN TERM 1

AS-91-651

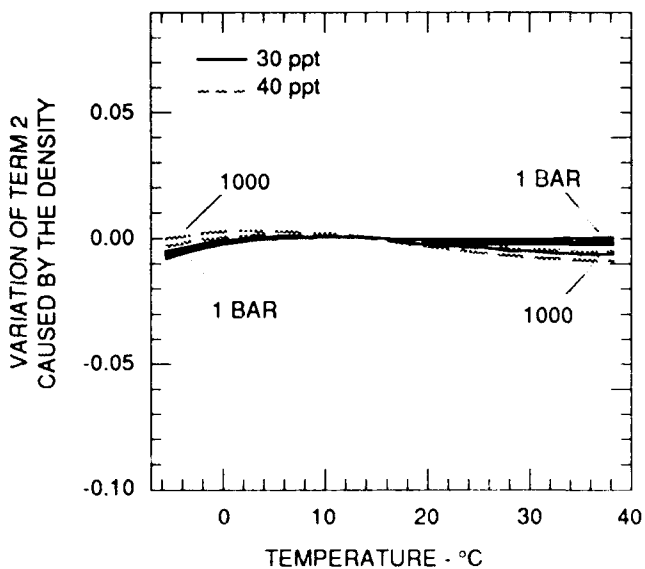


FIGURE 3.4
PLOTS OF THE ABSOLUTE DIFFERENCE IN β versus TEMPERATURE FOR ABSOLUTE PRESSURES OF 1, 100, 200, 300...800 BAR CAUSED BY USING DIFFERENT EMPIRICAL RELATIONS FOR ρ AND α IN TERM 2

AS-91-652

Rather than embark on an investigation of the origin of the above cited differences, we simply chose to use the relations from Chen, Millero, and their co-workers over those of other workers. Our reasoning was uniformity of approach: Chen, Millero, and their co-workers have developed empirical relations not only for c , but also for ρ , α , and C_P . Thus, for seawater, the relations in Refs. 10 and 12 were used for c and ρ . Estimates of α were obtained using the explicit relation for α in seawater from Ref. 12. When using our polynomial expressions to calculate β , however, one must always bear in mind that the differences cited above are, generally speaking, the accuracy limits in our results.

4. RESULTS

4.1 PLOTS OF c , ρ , β , AND Λ

Plots of c , ρ , β , and Λ for fresh water are shown in Figs. 4.1–4.4 and for seawater in Figs. 4.5–4.8. For fresh water, the acoustic properties are plotted as functions of temperature for the absolute pressures of 1, 100, 200, 300, ..., 1000 bars. Because the fresh water C_P relation is valid up to only 90°C, our relations for β and Λ in fresh water must be used with caution above 90°C. The curves are therefore dashed in this region. In the case of seawater, Figs. 4.5–4.8, the salinity is held constant at 35 ppt. The acoustic properties are plotted as functions of temperature for absolute pressures of 1, 100, 200, 300, ..., 800 bars. Because the accuracy of the relations for c and ρ in seawater is uncertain below 0°C, the values for c and ρ below 0°C are dashed lines in this region. Similarly, our derived relations for β and Λ in seawater must be used with caution below 0°C, and they too are dashed lines in this region.

Some discussion of the plots is in order. The plots of the sound speed and density for both fresh water and seawater are as expected: the density decreases with rising temperature, but increases with rising pressure, whereas the sound speed increases with pressure and, up to around 70–80°, increases with temperature also. The plots of the coefficient of nonlinearity reveal a more complex temperature and pressure dependence. In seawater, β increases with pressure for the temperature range considered. However, in fresh water, the sign of the pressure derivative of β is temperature dependent. Below about 60°C, β increases monotonically with pressure, whereas above that temperature, β initially decreases with rising pressure. For both fresh water and seawater, the pressure effect on β tends to decrease with rising temperature. The variation in β near 0°C is about 25%, whereas at higher temperatures the variation decreases to about 5%. In contrast with β , the pressure effect on Λ increases with temperature for both fresh water and seawater. Near 0°C, the variation is about 5%, but by 40°C the variation has increased to about 25%, and (in fresh water) by 90°C the variation has increased to about 40%. Aside from a deviation at low pressure and temperature, the temperature and pressure dependences of Λ for both seawater and fresh water are generally opposite those of the sound speed c . This is reasonable given that $c^{5/2}$ appears in the denominator of Λ .

A comparison of the isobar plots of β for fresh water and seawater (Fig. 4.3 with Fig. 4.7) casts some doubt on the validity of our extrapolation of the underlying expressions for the acoustic properties below 0°C. Note that, in our fresh water β

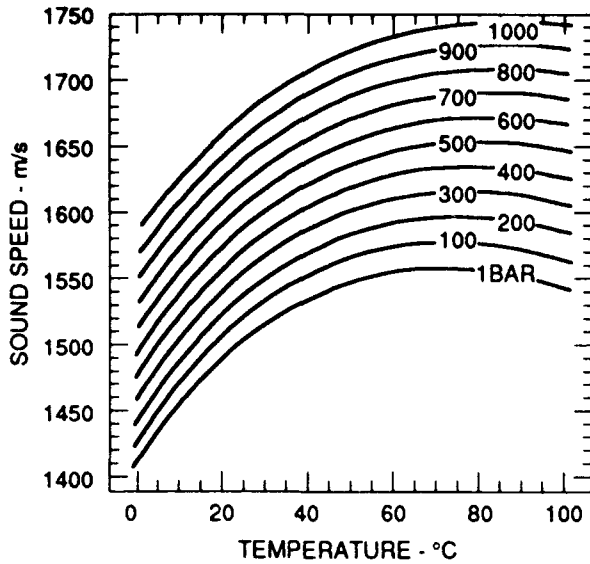


FIGURE 4.1
PLOTS OF SOUND SPEED versus TEMPERATURE
FOR VARIOUS ABSOLUTE PRESSURES (BAR)
IN FRESH WATER

AS-91-653

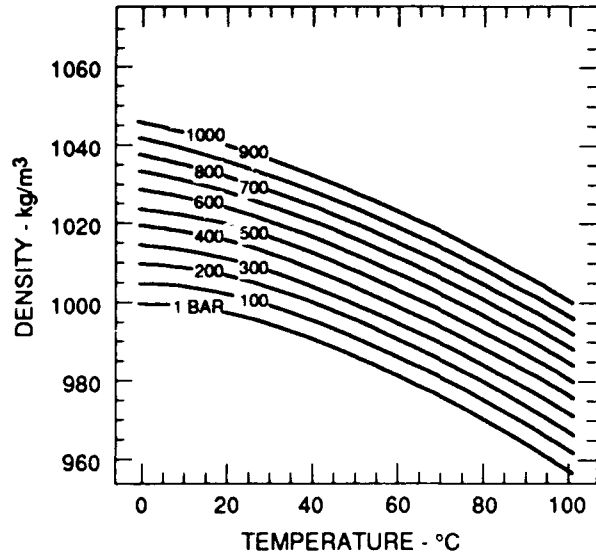


FIGURE 4.2
PLOTS OF DENSITY versus TEMPERATURE
FOR VARIOUS ABSOLUTE PRESSURES (BAR)
IN FRESH WATER

AS-91-654

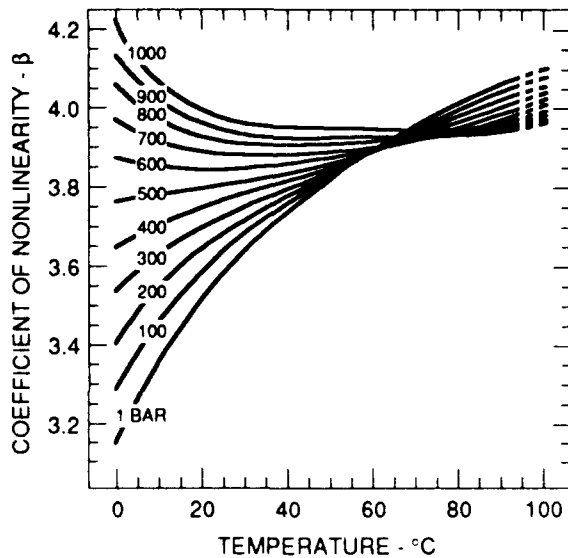


FIGURE 4.3
PLOTS OF β versus TEMPERATURE
FOR VARIOUS ABSOLUTE PRESSURES (BAR)
IN FRESH WATER

AS 91-655

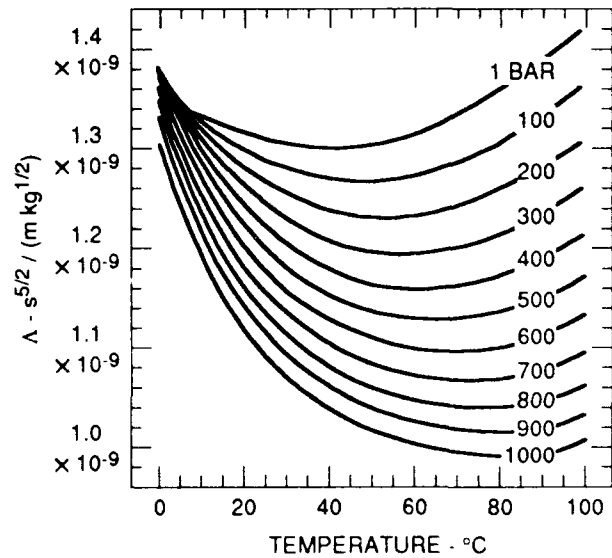


FIGURE 4.4
PLOTS OF λ versus TEMPERATURE
FOR VARIOUS ABSOLUTE PRESSURES (BAR)
IN FRESH WATER

AS 91 656

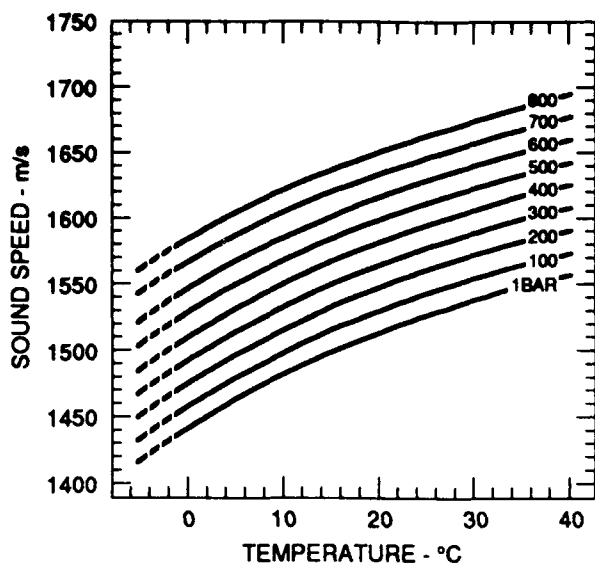


FIGURE 4.5
PLOTS OF SOUND SPEED versus TEMPERATURE
FOR VARIOUS ABSOLUTE PRESSURES (BAR)
IN SEAWATER WITH A SALINITY OF 35 ppt [10]

AS-91-657

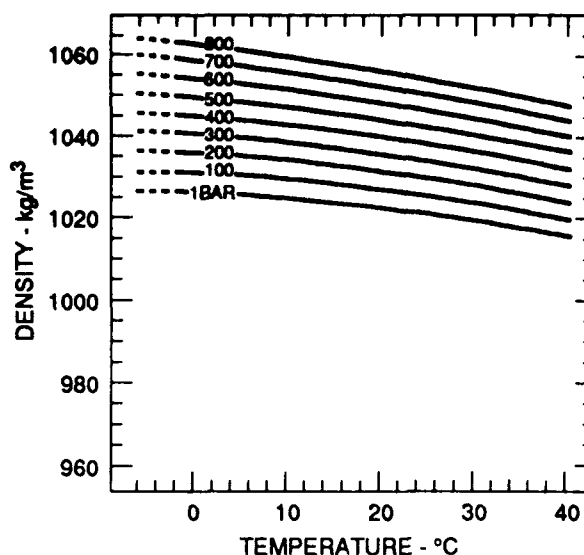


FIGURE 4.6
PLOTS OF DENSITY versus TEMPERATURE
FOR VARIOUS ABSOLUTE PRESSURES (BAR)
IN SEAWATER WITH A SALINITY OF 35 ppt [12]

AS-91-658

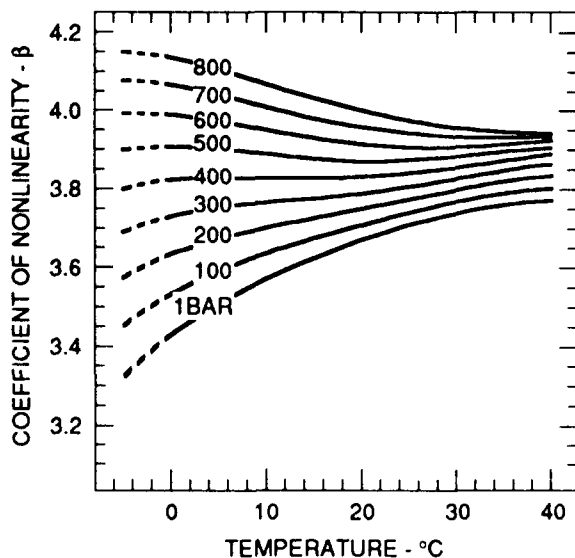


FIGURE 4.7
PLOTS OF β versus TEMPERATURE
FOR VARIOUS ABSOLUTE PRESSURES (BAR)
IN SEAWATER WITH A SALINITY OF 35 ppt

AS-91-659

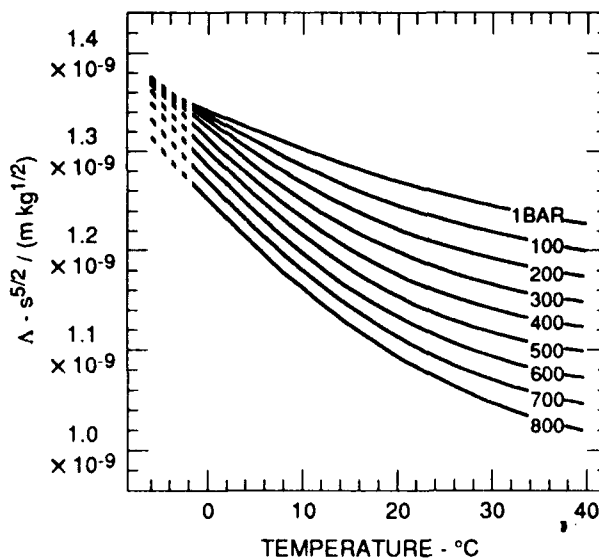


FIGURE 4.8
PLOTS OF Λ versus TEMPERATURE
FOR VARIOUS ABSOLUTE PRESSURES (BAR)
IN SEAWATER WITH A SALINITY OF 35 ppt

AS-91-660

results, the β curves diverge rapidly as they approach the freezing point, whereas in our seawater β results, the β curves do not exhibit a similar divergence below 0°C. The reason may be our unjustified extrapolation of the underlying expressions for c , ρ , α , and C_P in seawater. Although these relations are not stated as being accurate below 0°C, they are the best available. Because we believe that having an expression for β with an uncertain accuracy below 0°C is preferable over no expression at all, we extrapolated the underlying expressions to develop our relation for β .

4.2 POLYNOMIAL EXPRESSIONS FOR β AND Λ

Our polynomial expressions for β and Λ are now presented. The order of our polynomial fit was chosen to be similar to that used in the empirical fit to the dominant underlying acoustic property, the sound speed c . The sound speed relation given by Chen and Millero [7] is fifth order in temperature, third order in pressure, and second order in salinity. The order of the salinity terms was, however, reduced from second to first order because the sound speed relation given by Chen and Millero [7] is valid for a broader range of salinities than is required (5–40 ppt rather than our 30–40 ppt). The resulting fit of the polynomials to the original data is more than adequate. The quality of the fits of the polynomials to the original data was assessed through direct comparison, and the comparison was performed over the claimed ranges of temperatures, pressures, and salinities. The difference between the original data and the fit is always far less than the previously discussed uncertainty in the overall calculation.

Our polynomial expressions for β were obtained as follows. For seawater, Eq. (2.1) was evaluated over the following range: temperature -4 to 40°C , pressure 1–800 bars absolute, and salinity 30–40 ppt. The resulting data points were then fitted to a polynomial expression that is fifth order in temperature, third order in pressure, and linear in salinity. The polynomial expression is

$$\beta = \sum_{l,m,n} B_{lmn} (S - 35)^l P^m T^n, \quad (4.1)$$

where the salinity S is in ppt, the temperature T is in $^\circ\text{C}$, and the pressure P is bars gauge, that is, relative to 1 atm = 1.01325 bar.¹ The coefficients B_{lmn} are listed in Table 4.1. The maximum absolute error in the fit is 0.0015, and the typical error is about half that. (Check value: For a salinity of 32.5 ppt, a pressure of 100 bar gauge, and a temperature of 15°C , β is 3.672556.)

The actual salinity of the ocean water column is frequently unknown. In its place, people often assume a salinity of 35 ppt because this salinity is so common in the

¹This is the convention used by most authors, including Chen, Millero, and their co-workers, for presenting empirical fits to seawater and fresh water property data.

TABLE 4.1
Coefficients in polynomial expression for β

l	m	n	B_{mn}^0	B_{lmn}	B_{mn}^{35}
0	0	0	+3.122127	+3.44166	+3.44223
0	0	1	+0.0262768	+0.0176224	+0.0176264
0	0	2	-0.000439164	-0.000478735	-0.00047893
0	0	3	+0.00000520032	+0.0000129976	+0.0000130006
0	0	4	-3.07095×10^{-8}	-1.85398×10^{-7}	-1.85429×10^{-7}
0	0	5	$+6.12344 \times 10^{-11}$	$+3.36832 \times 10^{-10}$	$+3.37367 \times 10^{-10}$
0	1	0	+0.00139138	+0.00104125	+0.0010417
0	1	1	-0.000051128	-0.0000457645	-0.0000457763
0	1	2	$+7.40662 \times 10^{-7}$	$+5.84691 \times 10^{-7}$	$+5.85248 \times 10^{-7}$
0	1	3	-6.41226×10^{-9}	$+5.16276 \times 10^{-9}$	$+5.14004 \times 10^{-9}$
0	1	4	$+2.53274 \times 10^{-11}$	-3.84942×10^{-11}	-3.82208×10^{-11}
0	2	0	-2.15517×10^{-7}	-1.69183×10^{-7}	-1.69416×10^{-7}
0	2	1	$+3.45228 \times 10^{-9}$	$+1.69021 \times 10^{-8}$	$+1.69053 \times 10^{-8}$
0	2	2	$+1.44653 \times 10^{-10}$	-3.37325×10^{-10}	-3.37744×10^{-10}
0	2	3	-2.82686×10^{-12}	-3.82096×10^{-12}	-3.79902×10^{-12}
0	2	4	$+1.26217 \times 10^{-14}$	$+4.52996 \times 10^{-14}$	$+4.49897 \times 10^{-14}$
0	3	0	-1.12614×10^{-10}	-6.58146×10^{-11}	-6.57756×10^{-11}
0	3	1	$+4.51141 \times 10^{-12}$	$+6.60716 \times 10^{-12}$	$+6.60637 \times 10^{-12}$
0	3	2	-2.91434×10^{-14}	-1.63621×10^{-13}	-1.63501×10^{-13}
1	0	0		+0.00833572	
1	0	1		-0.000232407	
1	0	2		+0.00000107536	
1	1	0		-0.0000115765	
1	1	1		$+7.2387 \times 10^{-9}$	
1	1	2		$+1.0786 \times 10^{-8}$	
1	2	0		$+3.2689 \times 10^{-9}$	
1	2	1		$+4.6156 \times 10^{-10}$	
1	2	2		4.86744×10^{-11}	

world's oceans. For these reasons, a less complicated relation for β for the salinity of 35 ppt was also developed,

$$\beta^{35} = \sum_{m,n} B_{mn}^{35} P^m T^n \quad . \quad (4.2)$$

Valid over the same range of temperatures and pressures, its coefficients are denoted B_{mn}^{35} and are listed in Table 4.1. The maximum absolute error in the fit is 0.0001. (Check value: For a pressure of 98.98675 bar gauge and a temperature of 15°C, β^{35} is 3.683292.)

For fresh water, Eq. (2.1) was evaluated over the temperature range 0–100°C and the pressure range 1–1000 bars absolute. A polynomial fit of the same order in temperature and pressure as above was then obtained,

$$\beta^0 = \sum_{m,n} B_{mn}^0 P^m T^n \quad . \quad (4.3)$$

The coefficients B_{mn}^0 are also listed in Table 4.1, and the maximum absolute error of the fit is 0.0005. (Check value: For a pressure of 98.98675 bar gauge and a temperature of 15°C, β^0 is 3.508381.)

Polynomial fits for $\Lambda \equiv \beta/(\rho c^5)^{1/2}$ that are valid over the same ranges of temperatures and pressures for seawater with salinities of 30–40 ppt, seawater with a salinity of 35 ppt, and fresh water were developed as above. The relations are shown in Eqs. (4.4)–(4.6), and the coefficients, which are denoted, respectively, L_{lmn} , L_{lmn}^{35} , and L_{mn}^0 , are listed in Table 4.3:

$$\Lambda = \left(\sum_{l,m,n} L_{lmn} (S - 35)^l P^m T^n \right) \times 10^{-9} \quad \text{s}^{5/2} \text{m}^{-1} \text{kg}^{-1/2} \quad , \quad (4.4)$$

$$\Lambda^{35} = \left(\sum_{m,n} L_{mn}^{35} P^m T^n \right) \times 10^{-9} \quad \text{s}^{5/2} \text{m}^{-1} \text{kg}^{-1/2} \quad , \quad (4.5)$$

$$\Lambda^0 = \left(\sum_{m,n} L_{mn}^0 P^m T^n \right) \times 10^{-9} \quad \text{s}^{5/2} \text{m}^{-1} \text{kg}^{-1/2} \quad . \quad (4.6)$$

The maximum absolute errors in these fits are, respectively, 0.0004×10^{-9} , 0.0001×10^{-9} , and 0.0015×10^{-9} . (Check values: For a salinity of 32.5 ppt, a pressure of 100 bar gauge, and a temperature of 15°C, Λ is 1.270685×10^{-9} . For a pressure of 98.98675 bar gauge and a temperature of 15°C, Λ^{35} is 1.267513×10^{-9} and Λ^0 is 1.309151×10^{-9} .)

TABLE 4.2
Coefficients in polynomial expression for Λ .

l	m	n	L_{mn}^0	L_{lmn}	L_{mn}^{35}
0	0	0	+1.340649	+1.34255	+1.34282
0	0	1	-0.00123676	-0.00369707	-0.00370087
0	0	2	-0.0000219506	-0.0000454961	-0.0000454702
0	0	3	$+7.51912 \times 10^{-7}$	$+0.00000411388$	$+0.0000041155$
0	0	4	-2.82344×10^{-9}	-7.30211×10^{-8}	-7.30791×10^{-8}
0	0	5	-5.22167×10^{-12}	$+2.49154 \times 10^{-10}$	$+2.49717 \times 10^{-10}$
0	1	0	+0.000197704	-2.88818×10^{-7}	-4.62069×10^{-7}
0	1	1	-0.0000256345	-0.0000191962	-0.0000191921
0	1	2	$+4.70095 \times 10^{-7}$	$+4.28444 \times 10^{-7}$	$+4.28422 \times 10^{-7}$
0	1	3	-4.83161×10^{-9}	-4.12041×10^{-9}	-4.1215×10^{-9}
0	1	4	$+1.90758 \times 10^{-11}$	$+4.36867 \times 10^{-11}$	$+4.37169 \times 10^{-11}$
0	2	0	-2.69563×10^{-7}	-1.43851×10^{-7}	-1.43841×10^{-7}
0	2	1	$+1.46666 \times 10^{-8}$	$+1.65626 \times 10^{-8}$	$+1.65666 \times 10^{-8}$
0	2	2	-1.99597×10^{-10}	-3.94804×10^{-10}	-3.95058×10^{-10}
0	2	3	$+1.59563 \times 10^{-12}$	$+2.14739 \times 10^{-12}$	$+2.15271 \times 10^{-12}$
0	2	4	-5.86015×10^{-15}	-1.49805×10^{-14}	-1.50419×10^{-14}
0	3	0	$+4.01117 \times 10^{-11}$	$+2.48433 \times 10^{-11}$	$+2.48674 \times 10^{-11}$
0	3	1	-1.60177×10^{-12}	-2.5426×10^{-12}	-2.54489×10^{-12}
0	3	2	$+8.41646 \times 10^{-15}$	$+4.69605 \times 10^{-14}$	$+4.70247 \times 10^{-14}$
1	0	0		-0.000374879	
1	0	1		-0.0000614364	
1	0	2		$+4.68875 \times 10^{-7}$	
1	1	0		-0.00000518082	
1	1	1		$+1.53554 \times 10^{-7}$	
1	1	2		$+9.56226 \times 10^{-10}$	
1	2	0		$+2.92378 \times 10^{-9}$	
1	2	1		$+2.13857 \times 10^{-11}$	
1	2	2		-3.87438×10^{-12}	

4.3 DISCREPANCY WITH THE TABULATED VALUES OF ENDO [1]

A small discrepancy exists between our results and those of Endo [1]. The magnitude of the difference in terms of β is at most 0.15 at high temperature and pressure. The results for high pressure, low temperature as well as those for low pressure, high temperature appear to agree well. Thus a cross term appears to be the source of the problem. However, despite numerous tests and repeated checking, the source of the discrepancy could not be traced. The checks included evaluating the sound speed derivatives, Eq. (2.1), numerically for comparison with the values obtained via analytical differentiation. The agreement was within 1 ppt for term 1, and even less for term 2.

The discrepancy is particularly unsettling, as both our and Endo's underlying expressions for c , ρ , and α are the same. However, the discrepancy is small, of the same order as the uncertainty in β caused by using the different expressions for c , ρ , and α .

5. SUMMARY

In this work three polynomial relations for β are developed: one for fresh water as a function of temperature and pressure, one for seawater with a salinity of 35 ppt as a function of temperature and pressure, and one for seawater as a function of temperature, pressure, and salinity. Also developed are a similar set of relations for the term $\Lambda \equiv \beta/(\rho c^5)^{1/2}$. Plots of the sound speed c , the density ρ , as well as β and Λ for both fresh water and seawater with a salinity of 35 ppt are presented for a wide range of temperatures and pressures. Although the precision of the polynomial fits to the underlying data is quite good (typically better than 1 ppt), the use of different underlying property data leads to uncertainties in β of about $\pm 3\%$. Due to a lack of comparison information, a similar estimate of the uncertainty is not available for β in fresh water.

ACKNOWLEDGMENTS

It is a pleasure to acknowledge the financial assistance of the United States Office of Naval Research and the encouragement of the authors' host at Applied Research Laboratories, Dr. David T. Blackstock. The assistance of A. J. Kimbrough in numerically implementing the empirical relations and their derivatives is also greatly appreciated.

REFERENCES

- [1] H. Endo, "Calculation of nonlinearity parameter for seawater," *J. Acoust. Soc. Am.* **76**, 274-279 (1984).
- [2] L. A. Ostrovsky, E. N. Pelinovsky, and V. E. Fridman, "Propagation of finite-amplitude sound waves in an inhomogeneous medium with caustics," *Sov. Phys. Acoust.* **22**, 516-520 (1976).
- [3] C. L. Morfey, "Nonlinear Propagation in a Depth-Dependent Ocean," Applied Research Laboratories: The University of Texas at Austin, Tech. Rept. ARL-TR-84-11 (1984) (ADA 145 079).
- [4] I. Rudnick, "On the attenuation of finite amplitude waves in a liquid," *J. Acoust. Soc. Am.* **30**, 564-567 (1958).
- [5] R. T. Beyer, "Parameter of nonlinearity in fluids," *J. Acoust. Soc. Am.* **32**, 719-721 (1960).
- [6] Chen-Tung Chen and F. J. Millero, "Reevaluation of Wilson's sound-speed measurements for pure water," *J. Acoust. Soc. Am.* **60**, 1270-1273 (1976).
- [7] Chen-Tung Chen, R. A. Fine, and F. J. Millero, "The equation of state of pure water determined from sound speeds," *J. Chem. Phys.* **66**, 2142-2144 (1977).
- [8] Staff of ESDU International plc (London, UK), "Specific heat capacity at constant pressure of ice, water and steam," Engineering Sciences Data Unit Item No. 68008, amendment C (1977).
- [9] J. R. Lovett, "Merged seawater sound-speed equations," *J. Acoust. Soc. Am.* **63**, 1713-1718 (1978).
- [10] Chen-Tung Chen and F. J. Millero, "Speed of sound in seawater at high pressures," *J. Acoust. Soc. Am.* **62**, 1129-1135 (1977).
- [11] B. Gebhart and J. C. Mollendorf, "A new density relation for pure and saline water," *Deep-Sea Research* **24**, 831-848 (1977).
- [12] Chen-Tung Chen and F. J. Millero, "The specific volume of seawater at high pressures," *Deep-Sea Research* **23**, 595-612 (1976).

- [13] F. J. Millero, G. Perron, and J. E. Desnoyers, "Heat capacity of seawater solutions from 5° to 35°C and 0.5 to 22 ppt chlorinity," *J. Geophys. Res.* **78**, 4499–4507 (1973).
- [14] H. B. Callen, *Thermodynamics and an Introduction to Thermostatistics* (John Wiley and Sons, Inc., New York, 1985). See Problem 7.4–8.

5 November 1991

DISTRIBUTION LIST FOR
ARL-TR-91-27
UNDER ONR GRANT NO. N00014-89-J-1109

Copy No.

1-2	Office of Naval Research Physics Division Office (Code 1112) 800 North Quincy Street Arlington, VA 22217-5000
3	Naval Research Laboratory Department of the Navy (Code 2625) Washington, DC 20375-5000 Attn: Technical Library
4-5	Defense Technical Information Center Cameron Station Alexandria, VA 22314
6	Naval Underwater Systems Center New London, CT 06320 Attn: Technical Center
7	Naval Postgraduate School Monterey, CA 93940-5000 Attn: Technical Library (Code 0212)
8	Steven L. Garrett (Physics, Code PH/GX)
9	Anthony A. Atchley (Physics, Code PH/AY)
10	Commanding Officer Naval Oceanographic and Atmospheric Research Laboratory Stennis Space Center, MS 39529 Attn: Technical Library
11	Robert E. Apfel Yale University P.O. Box 2159 New Haven, CT 06520

Distribution List for ARL-TR-91-27 under Contract N00014-89-J-1109
(cont'd)

Copy No.

12	Yves H. Berthelot School of Mechanical Engineering Georgia Institute of Technology Atlanta, GA 30332
13	L. Bjørnø Technical University Denmark Department of Industrial Acoustics Building 425 DK-2800 Lyngby Denmark
14	M. A. Breazeale National Center for Physical Acoustics P.O. Box 847 Fraternity Row University. MS 38677
15	Franklin R. Breckenridge National Bureau of Standards Gaithersburg, MD 20899
16	M. J. Buckingham University of California Scripps Institution of Oceanography Marine Physical Laboratory La Jolla, CA 92093
17	Peter H. Ceperly George Mason University Electrical & Computer Engineering Dept. 4400 University Ave. Fairfax, VA 22030
18	Clarence S. Clay, Jr. 5033 St. Cyr. Rd. Middleton, WI 53562

Distribution List for ARL-TR-91-27 under Contract N00014-89-J-1109
(cont'd)

Copy No.

19-23 Frederick D. Cotaras
Defence Research Establishment Atlantic
P.O. Box 1012
Dartmouth, NS B2Y 3Z6
Canada

24 David G. Crighton
Cambridge University
Dept. of Applied Math. Theor. Phys.
Silver Street
Cambridge CB3 9EW
England

25 Lawrence A. Crum
National Center for Physical Acoustics
P.O. Box 847
University, MS 38677

26 H. Endo
The National Defense Academy
1-10-20 Hashirimizu
Yokosuka 239
Japan

27 J. H. Ginsburg
School of Mechanical Engineering
Georgia Institute of Technology
Atlanta, GA 30332

28 Robert E. Green
Materials Science and Engineering
The Johns Hopkins University
Baltimore, MD 21218

29 Victor Humphrey
School of Physics
University of Bath
Claverton Down
Bath BA2 7AY
United Kingdom

Distribution List for ARL-TR-91-27 under Contract N00014-89-J-1109
(cont'd)

Copy No.

30 Moises Levy
Department of Physics
The University of Wisconsin - Milwaukee
Milwaukee, WI 53201

31 Walter M. Madigosky
Naval Surface Warfare Ctr.
10901 New Hampshire Ave.
Silver Spring, MD 20903-5000

32 Philip L. Marston
Department of Physics
Washington State University
Pullman, WA 99164-2814

33 Walter G. Mayer
Department of Physics
Georgetown University
Washington, DC 20057

34 Julian D. Maynard
Department of Physics
Pennsylvania State University
University Park, PA 16802

35 Bryant E. McDonald
Naval Research Laboratory
P.O. Box 8337
Orlando, FL 32856

36 Herman Medwin
Ocean Acoustics Assoc.
4021 Sunridge Rd.
Pebble Beach, CA 93953

37-41 C. L. Morfey
ISVR
The University of Southampton
Southampton SO9 5NH
England

Distribution List for ARL-TR-91-27 under Contract N00014-89-J-1109
(cont'd)

Copy No.

42	Akira Nakamura Fukui Institute of Technology Electrical Engineering 6-1 Gakuen 3-Chrome Fukui 910 Japan
43	Allan D. Pierce The Pennsylvania State University Graduate Program in Acoustics University Park, PA 16802
44	Andrea Prosperetti The Johns Hopkins University Department of Mechanical Engineering Baltimore, MD 21218
45	Wolfgang H. Sachse Theoretical and Applied Mechanics Cornell University Ithaca, NY 14853-1503
46	University of Bergen Department of Mathematics Allegaten 55 5007 Bergen Norway
47	Attn: Sigve Tjøtta Jacqueline Naze Tjøtta
48	J. P. M. Trusler Imperial College London Department of Chemical Engineering Exhibition Road London SW7 2A7 United Kingdom

Distribution List for ARL-TR-91-27 under Contract N00014-89-J-1109
(cont'd)

Copy No.

49 James W. Wagner
Materials Science and Engineering
The Johns Hopkins University
Baltimore, MD 21218

U.T. Austin

50 A. Bedford (Engineering Mechanics)
51 I. Busch-Vishniac (Mechanical Engineering)
52 A. M. Gleeson (Physics)
53 T. A. Griffy (Physics)
54 M. F. Hamilton (Mechanical Engineering)
55 E. L. Hixson (Electrical & Computer Engineering)

ARL:UT

56 T. G. Muir
57 J. A. Shooter
58 P. Thorn
59 Library
60-65 NAD Reserve